

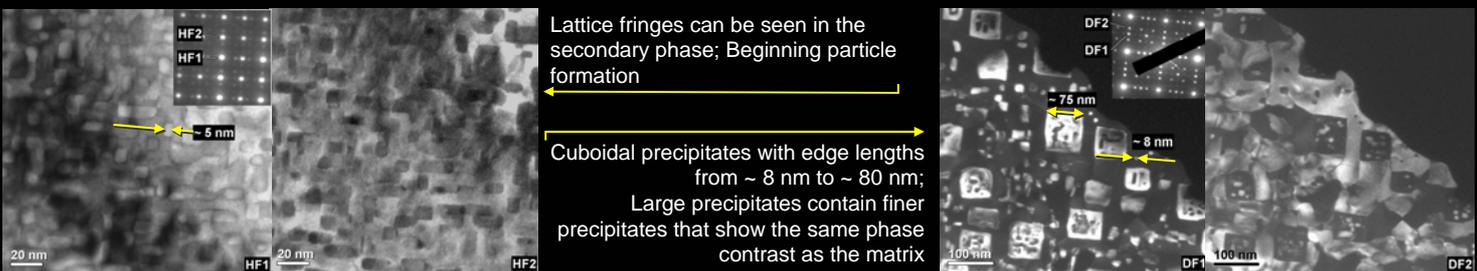
# Strengthening Mechanisms and Oxidation Behaviour in Platinum and Platinum Alloys

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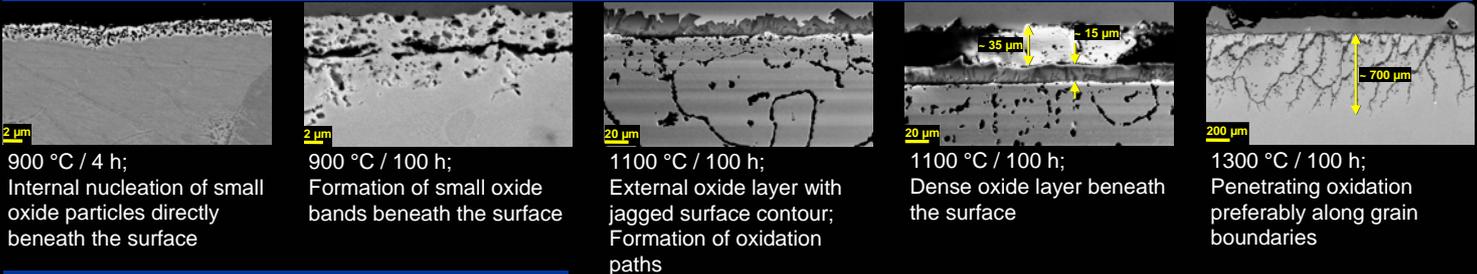
The chemical resistance against aggressive media even in the high temperature range and the catalytic properties qualify platinum as a material for a wide range of technical applications. Since the low mechanical strength of pure platinum is limiting its applicability, the use of suitable platinum alloys becomes indispensable. In binary systems precipitation hardening alloys have so far not been developed. Some ternary precipitation hardenable alloy systems that are reported in literature exhibit a high mechanical strength and seem to be promising candidates for technical use in the high temperature range. In contrast to the reputation of having an excellent oxidation resistance of Pt alloys it became evident that their oxidation behaviour is not sufficient yet and that degradation close to the surface of the alloys needs to be prevented.

## Bulk Material and Microstructure of the System Pt-Ti-Al

The present work shows Pt-Ti-Al-alloys that form ordered Pt<sub>3</sub>Ti phases and fulfil one of the prerequisites for technical applicability. In this work the Pt concentration is designed to be as high as possible (> 96 weight%). In spite of the low alloying element concentration the microstructures yield a hardness raise up to 500 HV (compared to 50 HV of pure Pt).



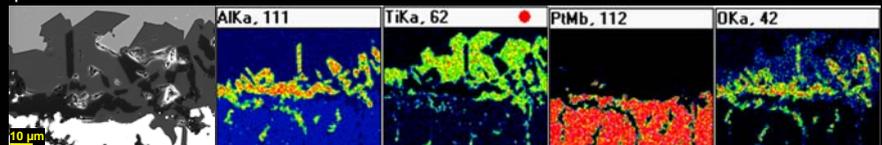
## Oxidation Behaviour in Air, Dependence on Temperature and Time



## Dependence on Al Concentration

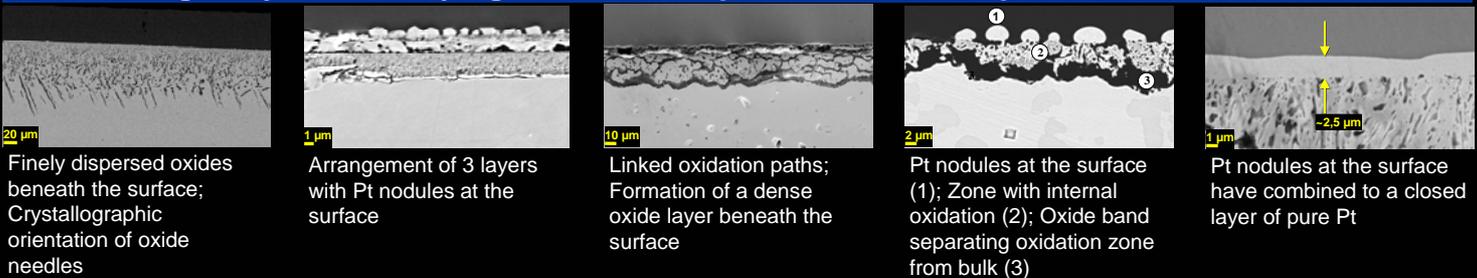


Binary Pt-Ti alloy: Intergranular oxidation only (1); Ternary Pt-Ti-Al alloy: Raising intragranular oxidation with increasing Al concentration (2), (3)

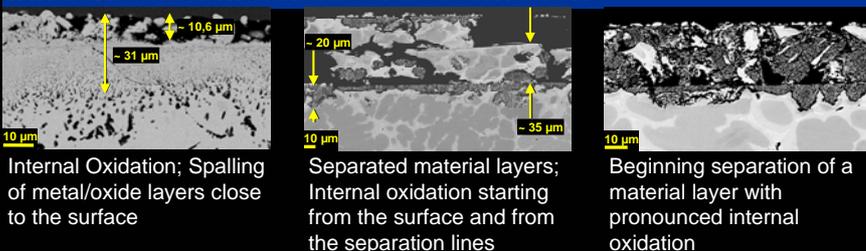


1100 °C / 100 h; Surface oxide layer: mixture of Al<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub>. The surface structure shows that outward diffusion of alloying elements is strongly contributing to the oxide formation. Element mapping: Al<sub>2</sub>O<sub>3</sub> forms directly at the surface as well as inside the bulk. TiO<sub>2</sub> forms on top of Al<sub>2</sub>O<sub>3</sub>. Some Al<sub>2</sub>O<sub>3</sub> particles are incorporated into the TiO<sub>2</sub> layer.

## Substituting Ti by other Alloying Elements, Dependence on Composition



## Influences of Mechanical Deformation



## Conclusions

A variety of morphologies of oxide particles and oxide layers can form on Pt alloys. Dense oxide layers are observed at or beneath the surface depending on composition, temperature and time. The morphology of the oxides can be influenced by different alloying elements. Mechanical deformations strongly change the oxidation behaviour and promote the formation of material layers that can show extensive spalling.